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FLAVONOID GLYCOSIDES FROM THE LEAVES OF *CISSUS IBUENSIS* HOOK (VITACEAE)

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## Abstract

The bioactive N-butanol fraction of the ethanol extract of the leaves of *Cissus ibuensis* was fractionated over silica Gel column to give Quercetin 3-O-rutinoside (I) and mixtures of Flavonoids (A2). A2 was fractionated using reverse phase HPLC to give Kaempferol 3-O- $\alpha$ -rhamnopyranosyl (1 $\rightarrow$ 6)- $\beta$ -D-galactopyranoside (II), Kaempferol 3-O-rutinoside(III)and Kaempferol3-O- $\alpha$ -rhamnopyranosyl(1 $\rightarrow$ 6)- $\alpha$ -rhamnopyranosyl(1 $\rightarrow$ 2)- $\beta$ -D-galactopyranoside (IV). The structures were elucidated by NMR spectroscopy and compared with literature.

**Key words:** Flavonoids; Kaempferol, Quercetin, anti-bacterial acitivity

## Introduction

*Cissus ibuensis* Hook (F) a climber belongs to the family Vitaceae and is distributed in the tropical regions particularly in Nigeria, Niger, Togo, Benin and Ghana. Traditionally the leaves of the plant are used in Northern Nigeria to treat gastrointestinal disturbance, as remedy for rheumatism and arthritis (Irvine, 1961)). The fruits and leaves decoction are applied as liniment for rheumatism and arthritis (Dalziel, 1958). While this species has not been investigated before. Investigation of various species from the genus *Cissus* have been described. The leaves of *C. rhifolia* contain quinolizidine alkaloids, flavonoids and terpenoids (Siafah et al., 1983). The stem wood of *C. pallida* showed the presence of Stilbenes, triterpenoids and steroids (Khan et al., 1986), while stilbenoids have been isolated from *C. quadrangularis* (Singh et al., 2005).

As part of our research focusing on the genus *Cissus* here we report the isolation of four Flavonoids from the N-butanol soluble part of the ethanol extract, we also report here the anti-bacterial activity of the Acetone, Ethanol and the partitioned fractions: Ethylacetate and N-butanol.

## Materials and methods

### General experimental procedure

Column chromatography was performed on silica gel G (60 - 120 $\mu$ m) BDH, TLC was performed on pre-coated Kieselgel 60 F<sub>254</sub> plates (Merck, Darmstadt, Germany); compounds were detected by spraying with Cerium sulphate in Sulphuric acid (Sigma-Aldrich). <sup>1</sup>H-NMR and <sup>13</sup>C-NMR were recorded on Bruker DRX spectrophotometer (300MHz) and 75MHz respectively in CD<sub>3</sub>OD with TMS as internal standard; Reversed phase (RP) HPLC separations were conducted on a waters 515 pumping system equipped with a Waters R401 refractive index detector, using a C<sub>18</sub> bondapak column (30cm x 7.8mm) and a mobile phase consisting of Methanol:Water(40:60) at flow rate of 2ml/min.

**Plant material:** The leaves of the plant *Cissus ibuensis* was collected from Samaru – Zaria in the month of June, 2004 and was identified by Mall.U. Gallah of the herbarium section, Department of Biological Sciences, Ahmadu Bello University, Zaria Nigeria where a voucher specimen No. 2708 was deposited.

**Extraction and isolation:** The air dried powdered leaves (250g) was extracted with 95% ethanol (1X2.5L to exhaustion by cold process. The combined ethanolic extract was concentrated at reduced pressure to afford a greenish mass (65g). 50g of this was suspended in water and filtered. The water soluble part was extracted with ethylacetate, and N-butanol. The n-butanol soluble part (2.0g) was packed with silica gel G (100g) in a column (50cm x 1.2cm) and eluted gradually with chloroform and chloroform : methanol mixtures. Progress of elution was monitored by silica TLC plates using n-butanol : Acetic acid : water (60 : 15 : 25) ethylacetate : methanol : water (100 : 16.5 : 13.5) and ethylacetate : formic acid : water (10 : 2 : 3). Elution with chloroform: methanol (8: 2) afforded a pale yellow solid which was crystallized in methanol to give compound I (10mg). Further elution with chloroform : methanol (8 : 2) gave A2(0.9g) which from TLC was shown to be mixtures of flavonoid. This was subjected to reverse-phase HPLC on a C<sub>18</sub>  $\mu$ -bondapak eluted with methanol : water (40 : 60) at 2ml/min to give compounds II(3mg), compound III(3.5mg) and compound IV(2.5mg) Compound IV was identified by <sup>1</sup>H-NMR, and 2D<sup>13</sup>C-NMR (HSQC, HMBC, TOCSY), while compounds I, II and III were identified by comparing their <sup>1</sup>H-NMR spectra with that reported in literature.

**Antibacterial studies:** The test organisms: *Staphylococcus aureus*, *Bacillus subtilis*, *Escherichia coli* and *Pseudomonas aeruginosa* are clinical isolates obtained from the Department of Pharmaceutical Microbiology and Biotechnology, Niger-Delta University, Wilberforce island Bayelsa state, Nigeria. The extracts at concentrations of 5 and 10mg/ml were screened for antibacterial activity by agar diffusion techniques as described by (Mendoza et al., 1997).

## Results

**Table 1:** <sup>1</sup>H- NMR of Compounds I-III (CD<sub>3</sub>OD) : 300MHz: J in Hz  $\delta$ ; in PPM

C	$\delta$ H	$\delta$ H	$\delta$ H
Aglycone moiety	I	II	III
2			
3			
4			
5			
6	6.15(d)	6.18(d),J=2.0	6.18(d),J=2.0
7			
8	6.33 (d)	6.37(d),J=2.0	6.37(d),J=2.0
9			
10			
1'			
2'	7.65( d), J=8.2	8.05(d),J=8.8	8.05(d),J=8.5
3'		6.86(d),J=8.8	6.89(d),J=8.5
4'			
5'	6.87 (d) J=8.4	6.86(d),J= 8.8	6.89(d),J=8.5
6'	7.58 (d,d) J=2.0	8.07(d),J= 8.8	8.05(d),J=8.5
$\beta$ -glucose/galactose			
1	5.58 (d),J=7.8	5.11(d),J=8.5	5.11(d),J=8.5
2	3.96		
3	3.72		
4	3.79		
5	3.64		
6	3.48		
rhamnosyl			
1	4.51(s)	4.51(s)	4.51(s)
2	4.03		
3	3.84		
4	3.37		
5	4.11		
6	1.02 (d), J=6.0	1.11(d),J=6.1	1.13(d),J=6.1

**Table 2:**  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR data of IV( $\text{CD}_3\text{OD}$ ) : 300/75MHz, J in Hz  $\delta$ ;in PPM)<sup>a</sup>

C	$\delta\text{H}$	$\delta\text{C}$
Aglycone moiety		
2		156.7
3		133.1
4		177.5
5		162.1
6	6.13(s)	101.5
7		164.3
8	6.19 (s)	96.0
9		156.7
10		104.0
1'		120.8
2'	8.06 d, J=8.2Hz	132.1
3'	6.90 d, J=8.2Hz	117.2
4'		160.1
5'	6.90 d, J=8.2Hz	117.2
6'	8.06 d, J=.2Hz	132.1
$\beta$ -galactosyl		
1	5.58 d,(J=7.8Hz)	100.6
2	3.96	77.5
3	3.72	75.5
4	3.79	70.6
5	3.64	75.2
6	3.48	66.8
rhamnosyl in (2)		
1	5.25(s)	102.3
2	4.03	72.2
3	3.84	72.1
4	3.37	73.9
5	4.11	69.8
6	1.02 d,(J=6.0Hz)	17.5
rhamnosyl in(6)		
1	4.55(s)	101.6
2	3.53	72.1
3	3.60	71.9
4	3.30	73.9
5	3.56	69.1
6	1.2 d,(J=6.0Hz)	17.6

a (assignments are based on HSQC, HMBC and TOCSY )experiments.

**Table 3:** Antibacterial activity of *Cissus ibuensis* extracts<sup>#</sup>

Tested material	<i>S. aureus</i>	<i>B. subtilis</i>	<i>E. coli</i>	<i>P. aeruginosa</i>
Acetone <sup>a</sup>	15.0	17.0	-	-
Ethanol <sup>a</sup>	15.0	18.0	15.0	15.0
Ethyl acetate <sup>a</sup>	18.0	15.0	13.0	15.0
N-Butanol <sup>a</sup>	18.0	15.0	15.0	16.0
Ciprofloxacin <sup>b</sup>	12.0	28.0	32.0	32.0
Gentamycin <sup>c</sup>	-	10.0	16.0	12.0

-: No inhibition

<sup>a</sup> Stock solution 10mg/ml

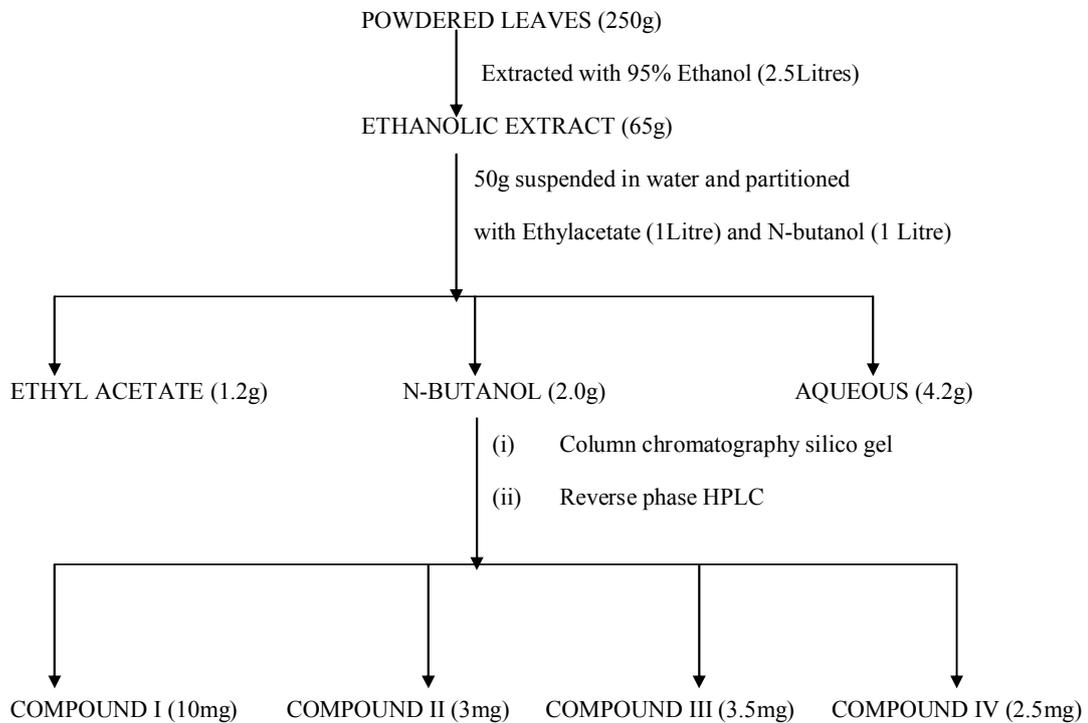
<sup>b</sup> 5 $\mu$ g/ml

<sup>c</sup> =10 $\mu$ g/ml

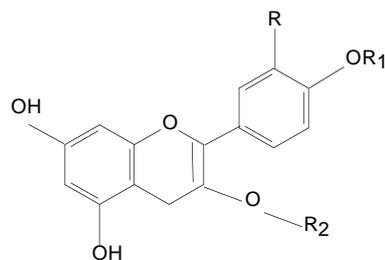
<sup>#</sup> Values are zone of inhibition diameter (mm) and mean of three replicates. The concentration of the standard antibiotic is 5 and 10 microgram microgram per disc.



**Figure 1:** HPLC Chromatogram of Fraction A2



**Figure 2:** Isolation scheme of the N-Butanol extract



I R=OH, R<sub>1</sub>=H, R<sub>2</sub>=rutinoside

II R=H, R<sub>1</sub>=H, R<sub>2</sub>= rhamnosyl(1→6)-galactoside

III R=H, R<sub>1</sub>=H, R<sub>2</sub>=rutinoside

IV R=H, R<sub>1</sub>=H, R<sub>2</sub>= rhamnosyl(1→2)rhamnosyl(1→6)- galactoside

Figure 3: Structures of isolated Flavonoids

## Discussion

Antibacterial studies of the extracts showed that the Ethanol extracts is more active than the acetone at concentration 10mg/ml against the test pathogens (Table 3) and the partitioned extracts, N-butanol showed slightly higher activity than the ethyl acetate fractions and this fraction gave the isolated flavonoids. Activity of the extracts against *S.aureus* is comparable to the standard antibiotic Ciprofloxacin while Gentamicin did not show any activity against *S.aureus*. Thus the extracts can be said to show more activity on *S.aureus* than the standard antibiotics used, however ciprofloxacin showed the best activity against the gram negative organisms used.

The <sup>1</sup>H-NMR spectrum of compound IV (Table 2) indicated a 5,7-dihydroxylated pattern for a ring A (two metacoupled protons at  $\delta=6.13$  and  $6.29$  ppm and an A<sub>2</sub>B<sub>2</sub> substitution pattern for ring B signals at  $\delta=6.91$  d, 2H, J=8 Hz and  $8.01$  d, J=8.0 Hz allowing the aglycon to be recognized as Kaempferol (Young Lin et al, 2001, Mabry et al., 1970). The <sup>1</sup>H-NMR spectrum of compound IV also showed signals ascribable to sugar moieties (Table 1), three anomeric protons arising from sugar moieties at  $\delta=4.55$  (s),  $5.25$  (s), and  $5.5$  d, J=(7.0 Hz) which correlated respectively with signals at  $\delta=102$  ppm,  $103$  ppm and  $101$  ppm in HSQC spectrum. All the <sup>1</sup>H and <sup>13</sup>C signals of compound IV (Table 2) were assigned using HSQC and HMBC experiments. Complete assignments of protons and carbon chemical shifts of the sugar portions were accomplished by 1D-TOCSY experiments and allowed the identification of the two terminal rhamnose units ( $\delta=1.04$ , d, 3H, J=6 Hz) and  $\delta=1.13$  d, 3H (J=6 Hz) and confirm by their corresponding carbon signals at  $\delta=17.2$  and  $18.1$  ppm respectively (Agrawal and Bansal, 1989).

The rhamnogalactosyl linkage was evident from the down field shift of the C-2 and C-6 of galactose sugar from  $71.3$  ppm to  $77.49$  ppm and  $60.8$  ppm to  $66.8$  ppm and up field shift of the C-1 and C-5 from  $102.0$  to  $100.4$  ppm and  $75.8$  to  $75.16$  ppm provided evidence that the linkage of the two rhamnose sugar is at C-2 and C-6 of the galactose (Markham et al., 1978), complete sugar assignment was aided by direct C-H correlation: HSQC, and TOCSY experiments. Thus compound IV was identified as Kaempferol-3-O- $\alpha$ -rhamnosyl-(1→6)- $\alpha$ -rhamnopyranosyl-(1→2)- $\beta$ -galactopyranoside. The <sup>1</sup>H-NMR spectra of compounds I, II and III (Table 1) were identified as Quercetin 3-O-rutinoside (rutin), Kaempferol 3-O- $\alpha$ -rhamnopyranosyl-(1→6)- $\beta$ -D-galactopyranoside and Kaempferol-3-O-rutinoside by comparison with that reported in literature (Yun-Lian et al., 2000; Ahmed and Nordin, 1998).

The presence of compounds I, II and III (Table 1) have been reported in leaf plants derived from Leguminosae and Moraceae (Luisa et al., 2000; Yun et al., 1991), but their presence in *Cissus ibuensis* is described for the first time. The isolated flavonoids might be responsible for the observed antibacterial activity as these plant constituents have been reported to have antibacterial actions (Harbone and Williams, 2000; VanPuyvelde et al., 1989). Work is ongoing to re-isolate these flavonoids and test their antibacterial activity.

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